

# Worksheet Balancing Equations Name I Fill In The

what is the atomic number of uranium topururanium has at no 92 and atomic weight 238 089 calculate the the atomic mass of uranium topurwhat is different isotopes of uranium number of protonsnumbers of calculate the energy released u 238 nucleus emits an alpha topurassume that the atomic mass of oxygen is topurthe binding energy per nucleon of uranium in mev is atomic mass of 9 define mass number and atomic number how are these topurthe binding energy per nucleon of uranium in mev is atomic mass of a uranium nucleus atomic number 92 mass number 238 emits an [www.bing.com](http://www.bing.com) what is the atomic number of uranium topur uranium has at no 92 and atomic weight 238 089 calculate the the atomic mass of uranium topur what is different isotopes of uranium number of protonsnumbers of calculate the energy released u 238 nucleus emits an alpha topur assume that the atomic mass of oxygen is topur the binding energy per nucleon of uranium in mev is atomic mass of 9 define mass number and atomic number how are these topur the binding energy per nucleon of uranium in mev is atomic mass of a uranium nucleus atomic number 92 mass number 238 emits an [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com) [www.bing.com](http://www.bing.com)

thus the atomic mass is 238 02891 g mol this is not to be confused with the mass number of 238 for the uranium 238 isotope which is merely an approximation to the isotopic mass of 238 0507882 20 u

the binding energy per nucleon of uranium in m ev is atomic mass of uranium ma 238 0508amu mass of hydrogen atom is m h 1 0078amu mass of neutron mn 1 0087amu atomic number of

the atomic mass of uranium 238 92 u is 238 0508 u while that of thorium 234 90 t h is 234 0436 u and that of helium 4 2h e is 4 0026 u alpha decay converts 238 92 u into 234 90 t h as shown

atomic masses are different for isotopes of uranium different isotopes of uranium have a different number of neutrons and different atomic masses they have the same atomic number the same

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assume that the atomic mass of oxygen is 7 a sample of 11 g of an oxide of uranium contains 10 g of uranium which of the following formula for the oxide is compatible with data

a uranium nucleus atomic number 92 mass number 238 emits an  $\alpha$  particle and the resultant nucleus emits  $\beta$  particle what are the atomic and mass numbers of the final nucleus

there are 15 protons and 16 neutrons in the nucleus of an element calculate its atomic number and mass number

an explosion of atomic bomb release an energy of 7 6 1013 j if 200 mev energy is released in fission of one u235 atom then the number of uranium atoms undergoing fission and the mass of uranium

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